1 Spatial variability of hydraulic conductivity and solute transport parameters and 2 their spatial correlations to soil properties

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11 Highlights

- We study reactive and non-reactive solute transport in a tropical soil.
- Solute transport parameters and hydraulic conductivity display high variability.
- Solute transport parameters and soil properties present spatial dependence.
- Spatial correlation is important up to 2.5 m.
- 16

17 Abstract

Spatial variation of the correlation among variables related to water flow and solute transport are important in the characterization of the spatial variability when performing uncertainty analysis and making uncertainty-qualified solute transport predictions. However, the spatial variation of the correlation between solute transport parameters and soil properties are rarely studied. In this study, the spatial correlation among laboratory-measured transport parameters dispersivity and coefficient of distribution of a reactive and a nonreactive solute and soil properties were studied at the scale of a few 25 meters using a dense sampling design. In an area of 84 m² and a depth of 2 meters, 55 26 undisturbed soil samples were taken to determine the soil properties. Column 27 experiments were performed, and the transport parameters were obtained by fitting the 28 experimental data to the analytical solution of the advection-dispersion equation using 29 the computer program CFITM. Stepwise multiple linear regression (MLR) was performed 30 in order to identify the statistically significant variables. The spatial correlation of the 31 variables and between variables were determined using the Stanford Geostatistical 32 Modeling Software. Soil properties presented a moderate coefficient of variation, while 33 hydraulic conductivity and transport parameters were widely dispersed. The difference 34 between its minimum and maximum value was guite large for most of the studied 35 variables evidencing their high variability. Both dispersivity and retardation factor were 36 higher than the expected and this result can be related to the preferential pathways and 37 to the non-connected micropores. None of the physical soil property was strongly 38 correlated to the transport parameters. Coefficient of distribution was strongly correlated 39 to the cation exchange capacity and significantly correlated to mesoporosity and 40 microporosity. Hydraulic conductivity presented significant positive correlation to the 41 effective porosity and macroporosity. Stepwise multiple linear regression analysis 42 indicated that further studies should be performed aiming to include other variables 43 relevant for lateritic soils such as pH, electrical conductivity, the content of Al and Fe, $CaCO_3$ and soil structure and microstructure. The study of the spatial correlation among 44 45 transport parameters and soil properties showed that the codispersion among the 46 variables is not constant in space and can be important in dictate the behavior of the combined variables. Our results also showed that some variables that were identified as 47

explanatory in the MLR were not significant in the spatial analysis of the correlation, showing the importance of this kind of analyses for a better decision about the most relevant variables and their relations. The present study was a first attempt to evaluate the spatial variation in the correlation coefficient of transport parameters of a reactive and a nonreactive solute, indicating the more relevant variables and the ones that should be included in future studies.

Keywords: cross-variogram, dispersivity, retardation factor, column experiment,
 undisturbed soil sample

56 **1.** Introduction

57 The soil's ability to retard and filter solutes as well as water flow and solute movement in 58 soils are significant themes in the earth and environmental sciences, and they are critical 59 in the hydrological and biogeochemical cycles (Keesstra et al., 2012; Kung et al., 2005). 60 Solutes can migrate from the soil to the groundwater and cause its contamination (Arias-61 Estévez et al., 2008). That ability can be guantified after determining soil transport parameters such as dispersivity (α) and partition coefficient (K_d) (Dyck et al., 2005; 62 63 Fetter, 1999). Knowledge of solute transport parameters is needed to improve the 64 prediction of the groundwater contamination potential (Kazemi et al., 2008). These parameters depend on many factors such as the chemical characteristics of the 65 66 contaminant and the soil physical, chemical, and physicochemical properties (Holland, 2004; Trangmar et al., 1986). 67

68 The transport parameters, the hydraulic conductivity, other soil properties and the 69 relations among them are highly spatially variable following a structural pattern 70 overlapped by an erratic component, also referred to as structured variation (Alletto and 71 Coquet, 2009; Fu and Gómez-Hernández, 2009; Goovaerts, 1997; Isaaks and 72 Srivastava, 1989; Mulla and Mc Bratney, 2002; Trangmar et al., 1986). The spatial 73 variability of soil properties might be studied at the centimeter scale, as well as at a 74 regional scale since the soil heterogeneity is present in all scales (Chapuis et al., 2005; 75 DeGroot and Baecher, 1993; Lacasse and Nadim, 1996; Søvik and Aagaard, 2003). 76 Additionally, since taking measurements of the properties of interest in an entire area is 77 impractical, there is always an uncertainty component related to the locations where the 78 properties were not measured (Erşahin et al., 2017; Fu and Gómez-Hernández, 2009).

79 The interest in quantifying the uncertainty in groundwater flow and solute transport 80 predictions has increased in the last decades (Cassiraga et al., 2005; Fu and Gómez-81 Hernández, 2009; Goovaerts, 2001; Grunwald et al., 2004; Hoffmann et al., 2014; 82 Lacasse and Nadim, 1996; Li et al., 2011; Teixeira et al., 2012). Performing an 83 uncertainty analysis and making uncertainty-gualified solute transport predictions 84 requires building a model of the spatial variability of the parameters controlling transport from a limited set of experimental data (laboratory or field). Such a model will allow 85 86 estimating soil properties at unsampled locations (Goovaerts, 1999).

87 The study of the spatial variability in soil science is commonly performed using geostatistics (Alletto and Coguet, 2009; Ersahin et al., 2017; Goovaerts, 1999; Gwenzi 88 89 et al., 2011; Marín-Castro et al., 2016). This technique is based on the random function 90 model assumption, where variables are modeled as random variables usually spatially 91 correlated. By assuming this model, the characterization of the spatial variability is 92 reduced to the characterization of the correlations among the random variables of the 93 random function (Goovaerts, 1997). Then, it is possible to perform coherent inferences 94 about the variable using estimation (such as kriging and cokriging) or simulation 95 techniques (such as sequential Gaussian simulation), and the spatial variability can be 96 fully characterized.

97 Geostatistics has been widely used to study the spatial variability of several soil 98 properties (Alletto and Coquet, 2009; Brocca et al., 2007; Goovaerts, 1998; Grego et al., 99 2006; Iqbal et al., 2005; Mbagwu, 1995; Tesfahunegn et al., 2011; Vieira, 1997; Wang 100 and Shao, 2013; Zhao et al., 2011) and specifically of the hydraulic conductivity (Bohling 101 et al., 2012; Gwenzi et al., 2011; Hu et al., 2008; Liu et al., 2017; Marín-Castro et al.,

2016; Motaghian and Mohammadi, 2011; Sobieraj et al., 2002; Sudicky et al., 2010). On
the other hand, the spatial characterization of solute transport parameters is still discrete
(Huysmans and Dassargues, 2006; Jacques et al., 1999; Kazemi et al., 2008) due to the
high cost and time-consuming efforts associated with solute transport studies (Erşahin et
al., 2017).

107 Allen-King et al. (2006) determined the spatial geostatistical properties of the 108 perchloroethene partition coefficient (K_d) and permeability (k) and found that K_d and k109 exhibited a statistically significant positive correlation. They concluded that additional 110 studies were necessary since the statistics describing the horizontal autocorrelation 111 behavior of *In* K_d and its cross-correlation to *In* k remained uncertain.

Gómez-Hernández, Fu, and Fernandez-Garcia (2006) studied the impact of the crosscorrelation between $ln K_d$ and ln K in the upscaling of the retardation factor (R) in a synthetic two-dimensional isotropic aquifer. They found that the upscaled R was highly affected by the cross-correlation between ln K and $ln K_d$. For a negative correlation, upscaled R for early times was smaller than that for late times. For a positive correlation, the result was the opposite and upscaled R for early times was larger than that for late times.

Erşahin et al. (2017) characterized the spatial variability of pore-water velocity (v), dispersivity, retardation factor and dispersion coefficient (D) and analyzed their statistical relations to other soil properties. They found that solute parameters were not correlated with the physical soil properties but were significantly correlated with soil chemical variables such as pH, electrical conductivity (*EC*) and cation exchange capacity (*CEC*).

124 A pure nugget model was fitted to $\log \alpha$ and R indicating no spatial structure. On the 125 contrary, $\log v$ and $\log D$ showed a moderate and strong spatial structure, respectively.

By analyzing many studies related to spatial variability in soil science, it can be noticed that a multivariate approach is used, in line with Goovaerts (1999), who points out that the soil information is generally multivariate. Usually, multivariate data are analyzed with statistical methods, such as principal component analysis or multiple linear regression (Ferreira da Silva et al., 2013; Rodríguez Martín et al., 2007) but without accounting for their possible spatial correlation (Erşahin et al., 2017; Kazemi et al., 2008). Ignoring the multivariate spatial correlations can be a waste of available and important information.

Some effort has been made to characterize the spatial variation of the correlation among variables and to use this information for estimation purposes (Benamghar and Gómez-Hernández, 2014; Bevington et al., 2016; Goovaerts, 1998; Guagliardi et al., 2013). Nevertheless, attempts to obtain the spatial variation of the correlation among solute transport and all statistically significant variables are rare (Jacques et al., 1999) and more studies need to be done.

139 Our first objective is to determine the linear statistical correlations among soils 140 properties, K, α , and K_d for a reactive (potassium) and a nonreactive (chloride) solute. 141 Second, in order to identify the more statistically significant variables that explain the 142 variability of the variables of interest (K, α , and K_{α}), multiple linear regression is 143 performed. The third objective is to model the spatial structures of soils properties and of 144 the variables of interest. Aiming to study the spatial cross-correlation among variables, 145 the fourth objective is to model the relations among the variables of interest and each 146 one of the more statistically significant variables. To the best of our knowledge, this is

147 the first time that the spatial correlations among α and K_d , of a reactive and a 148 nonreactive solute, and statistically significant variables are studied. Finally, although the 149 characterization of the spatial variability of soil properties at the centimeter/meter scale 150 can affect the solute transport prediction at a bigger scale (Salamon et al., 2007), 151 studies in this scale are scarce. In this context, we are interested in the small-scale 152 variability using a dense sampling design.

153 **2.** Material and methods

154 **2.1. Description of the study site**

155 The study was carried out in São Carlos city (21°51'38" S, 47°54'14" W), which is 156 located in the East-Center of the São Paulo State, Brazil (Fig.1). As mentioned before, 157 since we are studying the spatial variability at the scale of a few meters, the study site 158 covers an area of 84 m² and a depth of 2 meters. The pedologic soil type is classified as 159 Oxisol according to US Soil Taxonomy (Soil Survey Staff, 1999) and medium textured, 160 dystrophic, red-yellow Latosol according to the Brazilian classification system (Santos et 161 al., 2014). Clayey fine sand is the predominant texture. The climate in this region 162 is Köppen's Cwa type (Miranda et al., 2015; Peel et al., 2007). The mean annual 163 temperature is 21.2 °C, having humid and hot summers and a dry winter, with an 164 average annual rainfall of 1423 mm (Miranda et al., 2015). The parent material 165 comprising Cenozoic sediments that cover the Botucatu Formation (Paraná Sedimentary 166 Basin, São Bento Group), constituted by unconsolidated sands with the thickness 167 ranging from 5 to 7 m and pebbles at the base, and are spread at all São Paulo interior 168 region (Azevedo et al., 1981; Giacheti et al., 1993). The action of weathering under

tropical conditions makes the soil from the Cenozoic sediments highly lateritized (Giacheti et al., 1993). The main constituents of the studied soil are quartz, oxides, and hydroxides of aluminum, kaolinite, and gibbsite. Macropores and dual-porosity are also characteristics of that soil (Rohm, 1992).

173

174 Fig. 1

175 **2.2.** Soil sampling

176 Undisturbed soil samples were cautiously taken from hand-excavated trenches by 177 carefully forcing rigid polyvinyl chloride (PVC) cylinders (150 mm in height and 97.2 mm 178 in inner diameter) into the soil. Soil core sampling started by removing the grass (when 179 present) and a thin and hard layer from the top of the soil. Sampling in the x-y plane was 180 performed at 23 locations of the study site. For each x-y coordinate three samples were 181 taken at different depths (z coordinate 0.5 m, 1.0 m and 1.5 m), resulting in a dense 182 sampling design. Initially, 69 undisturbed soil samples were collected, but 14 samples 183 presented defects or cracks and were discarded. The position of the 55 remaining 184 samples in the study site is shown in Fig.1. Additionally, disturbed soil samples were 185 collected to characterize soil properties that were not spatially evaluated.

186 **2.3.** Soil properties characterization

Silt, clay and sand content, cation exchange capacity (*CEC*), total porosity (*n*), effective porosity (*ne*), macroporosity (*Ma*), mesoporosity (*Me*), microporosity (*Mi*) and bulk density (ρ_d) are referred to soil properties. These soil properties were analyzed spatially. In the laboratory, the moisture was determined in three replicates for each soil sample. Subsequently, the soil was air-dried and sieved through a #10 mesh sieve (2 mm openings). Particle size distribution were determined according to ASTM D7928-17 (ASTM, 2017a) and ASTM D6913 / D6913M-17 (ASTM, 2017b). in only one replicate for each soil sample. Particle density ρ_s was determined in five replicates using the ASTM D 854-14 (ASTM, 2014a) and resulted in 2.71 Mg·m⁻³ for all soil samples (arithmetic mean of all replicates).

Bulk density was determined for each soil column as $\rho_d = M_d/V_t$, where V_t is the total volume of the soil sample (internal volume of each PVC cylinder) and M_d is the dry mass of the soil sample.

200 Mercury intrusion porosimetry (MIP) (Washburn, 1921) and total porosity was calculated 201 for each soil sample as $n = 1 - \rho_d/\rho_s$. When the total porosity calculated was different 202 from the one obtained by MIP, we assumed that the difference was due to large pores 203 that were not identified in the MIP due to the reduced sample size used. The effective 204 porosity (ne) was considered as the total porosity minus the porosity that corresponds to 205 the soil water content at 33 kPa, suction equivalent to the field capacity (Ahuja et al., 206 1984; Brutsaert, 1967; Corey, 1977; Dippenaar, 2014). It is important to mention that the 207 field capacity is not precisely defined in soil science and we chose to use that value 208 since it is widely used in the literature. The diameter of the pore equivalent to the suction 209 at 33 kPa was calculated as 8.9 µm from the capillary rise equation assuming a contact 210 angle of 0°. Thus, based on the results of the MIP, the effective porosity was calculated 211 as the total porosity minus the porosity correspondent to the pores with a diameter smaller than 8.9 µm. From the MIP results, Ma, Me, and Mi were determined according 212 213 to the classification proposed by Koorevaar et al. (1983), in which the diameters of Mi,

Me, and *Ma* are, respectively, <30 μ m, 30-100 μ m and >100 μ m. The methylene blue adsorption test using the filter paper method described by Pejon (1992) was used to determine *CEC* in one replicate for each soil sample.

In order to characterize average properties with no concern about spatial structure, the next parameters were determined in three replicates by using disturbed soil samples: pH in H₂O and in KCl, redox potential (Eh) and electrical conductivity (*EC*) (Donagema and Campos, 2011), Δ pH (pH_{KCl} – pH_{H2O}) (Mekaru and Uehara, 1972), point of zero charge (PZC) (2pH_{KCl} – pH_{H2O}) (Keng and Uehara, 1974), organic matter content according to the ASTM D 2974-14 (ASTM, 2014b), and mineralogical composition by X-ray diffraction (Azaroff and Buerger, 1953).

224 **2.4.** Column experiments

225 The PVC cylinders used for collecting the undisturbed soil samples were used as rigid-226 wall permeameters and 55 column experiments were conducted. Figure 2 shows the 227 column experiments in progress. First, the columns were sealed with a cap containing a 228 stainless plastic plate with holes on both ends of the column, which allowed a uniform 229 distribution of the inlet flow. Second, the soil samples were slowly saturated from the 230 bottom with deionized water to remove entrapped air. Third, the flow was reversed, and 231 the permeability test was performed under a constant hydraulic gradient of 1 and the 232 flow rate (Q) was measured. We have taken two measures per day and we assumed 233 that steady-state flow was achieved when Q variations were below 5% in a week. Water 234 temperature was monitored throughout the test and, when necessary, corrections were 235 made in the calculations of the hydraulic parameters. Fourth, the following water flow 236 parameters were obtained from each soil sample: saturated hydraulic conductivity, K; 237 specific discharge, q; flow rate, Q; and average linear velocity, v (q/ne) (Freeze and 238 Cherry, 1979). When the steady-state flow was reached, deionized water was replaced by a 2.56 mol m⁻³ KCl solution (100 mg L⁻¹ K⁺ and 90.7 mg L⁻¹ Cl⁻ referred to as initial 239 240 concentrations, C_0 continuously injected into the soil column. Solute displacement tests 241 were carried out under constant hydraulic head and isothermal (20 °C) conditions. The 242 concentration, temperature, and pH of the initial solution were monitored throughout the 243 test. Leachate samples were collected from the outlet of the columns at pre-set time 244 intervals (defined for each column in accordance with the flow rate), stored 245 in plastic bottles and refrigerated immediately after collection. Preferably, the tests were 246 performed until the relative concentrations (C/C_0) reached 1, but this condition was not 247 achieved in some samples. An ion-selective electrode (ISE) (Hanna instruments - HI 248 4107 model) was used to determine Cl⁻ concentration (*C*) at each time. K⁺ concentration 249 at each time was measured by a flame photometer (Micronal B462 model) at a 1:21 250 ratio. All ion concentrations were measured in one replicate and determined as the 251 arithmetic mean of the replicates. The relative concentrations (C/C_0) of Cl⁻ and K⁺ were 252 determined by dividing the concentration of the ion in the leachate samples at each time 253 by the concentration of the ion in the initial solution. Thereafter, a breakthrough curve 254 (BTC) of each soil sample and each ion was plotted. The BTC's were expressed as C/C_0 255 and the number of pore volumes (T), where T is a dimensionless variable calculated as 256 T = vt/L (van Genuchten, 1980), where v is the average linear velocity, t is the time 257 elapsed from the start of the solute application, and L is the length of the soil column 258 (150 mm).

259

260 **Fig. 2**

261 **2.5.** Transport parameter determination

Dispersivity (α) [L] and partition coefficient between liquid and solid phases (K_d) [L³M⁻¹] are referred to as the transport parameters and were determined as explained next.

264 The advection-dispersion equation (ADE) used to interpret the BTCs is

265
$$R \frac{\partial C}{\partial t} = D \frac{\partial^2 C}{\partial x^2} - v \frac{\partial C}{\partial x}$$
, (1)

where C is solute concentration [ML-3], D is the hydrodynamic dispersion coefficient

[M2T-1], *R* is the retardation factor [-], *x* is the distance [L], and *t* is time [T].

268 The hydrodynamic dispersion coefficient is related to the dispersivity by

269
$$D = \alpha \cdot v$$
, (2)

and the retardation factor is related to the partition coefficient *Kd* through the expression $R = 1 + \frac{\rho_d}{p} K_d, \quad (3)$

This equation has the following analytical solution (Lapidus and Amundson, 1952; Ogata and Banks, 1961), when the initial condition is $C_0=0$ for the entire sample, and the boundary conditions are $C=C_0$ at the inlet and C=0 at an infinite distance from the inlet

275
$$\frac{C}{C_0} = \frac{1}{2} \left[\operatorname{erfc} \left(\frac{RL - vt}{2\sqrt{DRt}} \right) \right] + \frac{1}{2} \exp \left(\frac{vL}{D} \right) \operatorname{erfc} \left(\frac{RL + vt}{2\sqrt{DRt}} \right), \quad (4)$$

where erfc is the complementary error function.

This expression was fitted to the observed BTCs for each soil sample and values of *D* and *R* were obtained for both K⁺ and Cl⁻. The fitting was performed using the computer program CFITM (van Genuchten, 1980), that is part of the Windows-based computer software package Studio of Analytical Models (STANMOD) (Šimůnek, van Genuchten, Šejna, Toride, & Leij, 1999).

282 **2.6.** Statistical analysis

283 Exploratory analysis of the K, In K, soil properties and transport parameters (including P, 284 *R*, *D*, *In* α of K⁺ and *In* α of Cl⁻) was performed. Global summary statistics such as mean, 285 standard deviation, variance, minimum and maximum value, kurtosis, skewness, and 286 coefficient of variation (CV) were computed. The CV were classified according to Wilding and Drees (1983): low variability for CV < 15%; moderate variability for 15% < CV < 287 288 35%; and high variability for CV > 35%. The normality of the data was tested by means 289 of the Kolmogorov-Smirnov test (Massey, 1951). When necessary, the variables were 290 standardized, and the subsequent studies were performed using standard normal 291 variables. Outliers were analyzed in detail to investigate errors in the determination of 292 the variables, and when deemed appropriate they were excluded from the dataset. 293 Trends were also investigated and, if present, removed. The presence and strength of 294 significantly linear associations between soil properties and the variables of interest (In 295 K, In α (K⁺), In α (Cl⁻), Kd (K⁺), and Kd (Cl⁻)) were examined by computing Pearson 296 correlation coefficients at 0.05 and 0.01 significance level. The natural logarithm (In) of α 297 and K were used as variables rather α and K because they resulted in better 298 correlations.

The quantification of the significance of the relationships between all the studied variables, i.e. soil properties and variables of interest, was analyzed separately using multiple linear regression (MLR) (Eq. 3). Stepwise regression analyses were carried out to avoid the possible collinearity effects in multiple regressions. Statistically significant differences were set with p values equal to 0.05.

304 A stepwise MLR as in Equation (3)

305
$$y = b_0 + b_1 w_1 + b_2 w_2 + b_3 w_3 + \dots + b_n w_n$$
 (3)

306 defines the best linear combination of the variables to predict the variables of interest 307 and helps understand which variables have the highest influence on the variables of 308 interest,

309 where *y* is the dependent variable and w_1 to w_n are independent variables.

310 **2.7. Geostatistical analysis**

Based on the MLR results, the spatial dependence of the more statistically significant soil properties and the variables of interest was measured using direct experimental variograms.

The variogram can be defined as the mean-squared difference between the same variable at specified separation distances (Isaaks and Srivastava, 1989), and it was calculated using

317
$$\gamma(\mathbf{h}) = \frac{1}{2N} \sum_{aa=1}^{N(\mathbf{h})} [z(u_{aa}) - z(u_{aa} + \mathbf{h})]^2$$
 (5)

M(1)

where $\gamma(\mathbf{h})$ is the variogram function, $z(\mathbf{u}_{\alpha})$ is the measured value of the attribute under consideration taken at location *aa*, **h** is the separation vector and $N(\mathbf{h})$ is the number of data-pairs separated by the vector **h**. The variograms were obtained using the Stanford Geostatistical Modeling Software (SGeMS).

Almost all experimental variograms were best fitted to the isotropic spherical variogram
 model (Isaaks and Srivastava, 1989)

324
$$\gamma(\mathbf{h}) = c_0 + c_1 sph(|\mathbf{h}|, a)$$
 (6)

where *a* is the range, i.e., is the separation distance beyond which observations are spatially independent of each other, c_0 is the nugget effect, c_1 is the covariance contribution or sill value, and **h** is the directional lag distance.

The nugget effect model was also used in a situation, indicating that the variable was randomly spatially distributed.

330
$$\gamma(\mathbf{h}) = \begin{cases} 0 & if \ \mathbf{h} = 0 \\ 1 & otherwise \end{cases}$$
 (7)

In multivariate geostatistics, to model the coregionalization between p variables requires modeling p (p+1)/2 direct and cross-variograms. In this paper, p corresponds to the variables of interest plus the set of variables that best explains its variability, according to the MLR results.

The cross-variogram function describes the way in which two variables are spatially related, and was used to quantify the structure of the spatial correlation between selected soil properties and *In K*, *In α* (K⁺), *In α* (Cl⁻), K_d (K⁺), and K_d (Cl⁻¹)

338
$$\gamma_{uv}(\mathbf{h}) = \frac{1}{2N(\mathbf{h})} \sum_{i=1}^{N(\mathbf{h})} [z_i(u_{aa}) - z_i(u_{aa} + \mathbf{h})] \cdot [z_j(u_{aa}) - z_j(u_{aa} + \mathbf{h})]$$
(8)

where $z_i(u_{aa})$ and $z_j(u_{aa})$ are the measured z_i and z_j regionalized variables, respectively, taken at location *aa*.

The codispersion coefficient, r_{uv} , between the variables v and u, r_{uv} , for each vector **h** was computed for any pair of variables as the ratio of the cross-variogram between vand u to the square root of the product of the variograms of u and v (Goovaerts, 1994)

344
$$r_{uv}(\mathbf{h}) = \frac{\gamma_{uv}(\mathbf{h})}{\sqrt{\gamma_u(\mathbf{h})\gamma_v(\mathbf{h})}}$$
 (9)

345 **3. Results and discussion**

346 **3.1.** Average soil properties

347 The main minerals present in the studied soil are quartz, kaolinite, and gibbsite, in 348 accordance with Giachetti et al. (1993) and Kronberg et al. (1979). Average values of 349 5.71 and 5.19 for pH in H₂O and in KCI, were obtained, respectively. These results show 350 that the soil is strongly acid, which is a typical characteristic of Cenozoic sediments and 351 lateritic soils (Fagundes and Zuguette, 2011; Giacheti et al., 1993). The negative ΔpH (-0.52) and a point of zero charge (PZC) (4.67) lower than the pH_{H2O} indicate a 352 353 predominance of negative charges, which can promote cation adsorption (Fagundes and 354 Zuquette, 2011). This soil contains a small average amount of organic matter (2.40 %), a 355 result suitable for lateritic acid soils (Mahapatra et al., 1985). According to the soil 356 salinity classification of the Food and Agriculture Organization of the United Nations 357 (FAO), the electrical conductivity values indicate small amounts of dissolved salts (55.70 358 mS m⁻¹) and a non-saline soil (Abrol et al., 1988).

359 3.2. Soil properties statistical analysis

The exploratory statistical results of the soil properties, *v* and *K* are shown in Table 1. In order to identify trends, all statistical results were also investigated for each depth (results not shown), and no significant influence of the depth was observed. Because of that, in further analysis the samples were considered as a unique dataset, regardless of the depth. Soil properties are slightly skewed, quantified by a skewness < [0.5] (Webster, 2001), except *Ma* and *CEC*, which are moderately and highly skewed with a skewness of 0.75 and 1.06, respectively. The difference between its minimum and maximum value was quite large for K, *In K*, *v*, silt content, *Ma*, and *CEC*. According to the CV
classification of Wilding and Drees (1983), high CV were identified for *K*, *v*, silt content
and *Ma* (1.22, 1.23, 0.61 and 0.56, respectively) evidencing high variability in these
variables. Our results confirm that soil heterogeneity is present even on a small scale,
depending on the studied property (Chapuis et al., 2005; Lacasse & Nadim, 1996; Søvik
& Aagaard, 2003).

373 Mercury intrusion porosimetry results indicated that the soil has dual-porosity and the 374 predominant pore diameters correspond to Me and Mi. The multimodal pore size 375 distribution is characteristic of well-structured soils (Hajnos, Lipiec, Świeboda, 376 Sokołowska, & Witkowska-Walczak, 2006; Lipiec et al., 2007). The soil has a low CEC (maximum value 4.20 cmolc Kg⁻¹) and it suggests a low capacity to adsorb cations by 377 378 electrostatic adsorption (Fagundes & Zuguette, 2011). Mean soil properties presented 379 values in accordance with the typical characteristics of the studied soil (Giacheti et al., 380 1993; Zuquette & Palma, 2006), and are shown in Table1.

381

Table 1

383 3.3. Statistical analysis of the transport parameters

The breakthrough curves (not shown) of K⁺ and Cl⁻ obtained from the 55 miscible displacement tests were analyzed, and transport parameters were determined. The goodness of fit of the experimental BTC to the ADE model was evaluated by its R². Most BTCs presented significant tailing, R² ranged from 0.77 to 0.99 with a mean of 0.92 for K⁺ and 0.95 for Cl⁻, suggesting that the ADE model was suitable to describe the data. BTCs that presented low R² were investigated to check for problems in the soil samples,
but no problems were found.

Basic statistics of the transport parameters are shown in Table 2. Almost all transport parameters were high right-skewed. Moderate right-skewness was obtained only for *R* (Cl⁻) and K_d (Cl⁻). Slightly right-skewness was obtained for $ln \alpha$ (K⁺) and $ln \alpha$ (Cl⁻). High right-skewness bromide (Br⁻) α and *D*, and moderate $ln \alpha$ left-skewness was found in the work of Ersahin et al. (2017).

396 All transport parameters show high CVs and the highest ones were obtained for the 397 reactive solute (K⁺). The coefficients of variation of R and K_d for K⁺ shown that transport 398 parameters are very variable. The values we obtained for α were high when compared 399 to other studies using samples of approximately the same dimensions (Ersahin et al., 400 2017). Also, mean α values were high when compared to the typical values used in the 401 literature ($\alpha = 0.1L$, where L in the distance) (Freeze & Cherry, 1979). These differences 402 can be attributed to numerous factors such as the scale of the experiment, flow rate, and 403 boundary conditions. Higher values of α can also be indicative of preferential flow.

The maximum and minimum values were quite different for all transport parameters, evidencing, again, the large variability in these parameters. Peclet numbers ranged from 0.11 to 13.41, showing that for some soil samples the advective transport prevailed, whereas for other samples, dispersive transport was the primary mechanism. These differences probably are related to heterogeneities between physical characteristics of soil samples. *R* (K⁺) ranged from 0.69 to 36.19, while *R* (Cl⁻) ranged from 0.33 to 5.20, as expected because reactive solute should have larger *R* values than nonreactive solutes.

411 Even though clay content was significant, high R (K⁺) and R (Cl⁻) values were not 412 expected since the combination of the clay minerals identified, the low CEC values and 413 the predominance of negative charges do not favor the retardation of K⁺ and Cl⁻. We 414 believe that the structure of the soil played an essential role on the retardation. 415 Moreover, the results of P and R can be explained by the distribution of the diameter of 416 the pores in the soil, since the maximum Ma and Mi values were 0.15 and 0.36, 417 respectively. Because of that, part of the solutes can move fast because of advection (in 418 macropores) and part of them can be retarded due to the percolation through 419 micropores and non-interconnected pores, behavior also stated by others (Jarvis, 2007; 420 Silva, van Lier, Correa, Miranda, & Oliveira, 2016; van Genuchten & Wierenga, 1976).

421

422 **Table 2**

423 **3.4.** Correlation among variables

424 To examine the relationship among soil properties, hydraulic conductivity and transport 425 parameters, correlation coefficients were computed. Outliers were removed before the coefficients were computed and the analyses were performed using 50 values for each 426 427 variable. As none of the variables was normally distributed, correlation analyses were 428 performed using the original data (results not shown) as well as the standardized normal 429 distributed transformed values. As the best correlation coefficients were obtained with 430 standardized variables, all analyzes hereafter were performed using these variables. 431 Variables that are not intrinsic properties of the media such as P, D, R, and v, were not 432 considered in the analysis of correlations.

None of the physical soil property was strongly correlated to the transport parameters. According to Vanderborght and Vereecken (2007), texture has no significant effect in α and this result is also verified in our study. Since the studied soil has a structure characteristic of lateritic soils by forming agglomerates, texture itself may not show much about dispersivity.

It was obtained a statistically significant positive correlation between $ln \alpha$ (Cl⁻) and ρ_d and a negative correlation with *n*. This result is in accordance with the equation that relates dispersivity to *D* and *v* (*D* = αv , where *v* = q/ne). Since *n* is slightly negatively related to *ne*, as shown in Table 3, when *v* increases α decreases, justifying the relations obtained. The only variable significantly positively correlated to $ln \alpha$ (K⁺) was $ln \alpha$ (Cl⁻), suggesting that higher *D* smaller the influence of other soil properties.

A strong positive correlation was obtained between K_d (K⁺) and *CEC* and K_d (Cl⁻), showing the importance of the physico-chemical adsorption and the relation between the ions studied. A low, but still significant, positive correlation among K_d (K⁺) and *Me* was obtained. A negative correlation was obtained between K_d (K⁺) and *Mi*, indicating that neither *Ma* nor *Mi* contributed to higher *R*, contrary to our initial assumptions. A low positive correlation was presented by K_d (Cl⁺) with silt content and a strong positive correlation with *CEC* and K_d (K⁺) and no correlation with pore size was observed.

Almost no correlation was obtained among *CEC* and clay content, indicating that the clay mineral present in the soil is not relevant to adsorb cations, as mentioned before. Significant positive correlations among *In K*, *n*, *ne*, and *Ma* were verified, indicating that these properties dictate the values of *In K* and of the water flow in soils (Biswas & Si, 2009). In a previous study, a high positive correlation was obtained among *K*, *Ma*, and *n* 456 (Mbagwu, 1995). A significant negative correlation was also found among ln K and ρ_{d} , 457 results in accordance with other studies (Bevington et al., 2016; Botros, Harter, Onsoy, 458 Tuli, & Hopmans, 2009; Mbagwu, 1995; Papanicolaou et al., 2015). These results show 459 the higher ne (negatively related to n as shown in Error! Reference source not 460 found.), higher v, as expected. No significant correlation between In K and texture was 461 obtained. However, this result contrast with several previous studies in non-lateritic soils, 462 showing the impact of the soil agglomerates in the relation among soil properties (M. 463 Huang, Zettl, Lee Barbour, & Pratt, 2016; Igwe, 2005; Nemes, Timlin, Pachepsky, & 464 Rawls, 2009; Pachepsky & Rawls, 2004; Søvik & Aagaard, 2003).

465

466 **Table 3**

467

468 Table 4 presents the results of the stepwise multiple linear regression analysis at a 469 significance level of 95%. This analysis was used for investigating the significance of the 470 relationships among all selected variables. The best model for K_d (K⁺) was obtained by 471 considering two variables, CEC and Ma, explaining 70% of the total variance in the 472 model, with Pearson coefficient r equal to 0.84. The model that best represents K_d (Cl⁻) 473 was found by combining CEC, clay content and Me, which explain 60% of the total 474 variability with a moderate r equal to 0.70. These results suggest that other variables 475 that were not considered in this study could be added to better explain the total 476 variability of K_d . For example, several authors have suggested that pH, EC, the content 477 of AI and Fe, $CaCO_3$ and organic carbon have a strong influence on the total variability 478 of K_d (Che, Loux, Traina, & Logan, 1992; Erşahin et al., 2017; Porfiri, Montoya,

Koskinen, & Azcarate, 2015). Additionally, some variables that were significantly correlated to K_d in the correlation analysis were not significant in the MLR. This can be related to possible collinearity effects of these variables, what is identified and excluded by using stepwise method.

The only variables that were significant to model *ln* α (K⁺) and *ln* α (Cl⁻) were *ln* α (Cl⁻) and *ln* α (K⁺), respectively, and both have explained only 50% of the total variability, with a moderate *r* equal to 0.70 and 0.72, respectively. It demonstrates that other variables should be considered to better explain total variability in *ln* α . As α has some scale and spatial dependence (Erşahin et al., 2017; Freeze & Cherry, 1979), it could be interesting to take into account its spatial relationship with other parameters and not only the parameter itself.

490 Only 40% of the total variability of In K was explained by the combination of ne, ρ_d , and 491 Ma with a moderate r equal to 0.63. Contrary to the correlation analysis, where the 492 correlation between n and In K was statistically significant, in the MLR, n had not 493 explained In K variability when combined with other variables. In future studies, it would 494 be valuable to include other explanatory variables, such as soil structure and 495 microstructure that in previous studies were recognized as direct drivers of K (Benegas, 496 Ilstedt, Roupsard, Jones, & Malmer, 2014; Beven & Germann, 2013; Burke, Mulligan, & 497 Thornes, 1999; Hillel, 2004; Nanzyo, Shoji, & Dahlgren, 1993; Narwal, 2002; 498 Zimmermann & Elsenbeer, 2008) and can be even more important for lateritic soils.

499

500 **Table 4**

501 **3.5.** Spatial correlation among variables

502 As the correlation among variables in Table 3 neglects the spatial component of the 503 sample points, in this section, the direct and the cross-variograms are used to explore 504 further the spatial correlation among variables. The spatial structure of the standardized 505 variables was evaluated using variograms functions. Table 5 summarizes the 506 parameters of the models that were used to fit the experimental variograms. Only clay 507 content shows no spatial dependence (pure nugget effect), indicating that this variable is 508 spatially random, despite being correlated to sand and silt content, which display spatial 509 dependence. This result can be related to the more or less uniform distribution of the 510 clay content in the studied site, with a CV of only 9%. Experimental variograms of all the 511 remaining variables were fitted with a spherical model, indicating that abrupt changes in 512 space may occur, while preserving an overall spatial structure.

513 The spatial structure was similar for all the studied variables. The largest range was 514 obtained for In K (4.0 m), while silt content and Mi presented the smallest ones (2.5 m). 515 Microporosity, as well as all studied solute transport parameters, displayed a nugget 516 effect behavior, which accounts for short-scale spatial variability or measurement errors. 517 These variables had a moderate spatial dependence classified by measuring the nugget 518 ratio (R_b = nugget/sill*100%), which is strong if R_b < 25%, moderate if 25% < R_b < 75%, 519 weak if $R_b > 75\%$ (Cambardella et al., 1994). Variograms of K_d resulted in a greater 520 range than In α variograms. Gupte et al. (1996), found a maximum range of 2.3 m for Br⁻ 521 dispersivity. Contrary Erşahin et al. (2017) reported no clear spatial structure for α and R 522 under their sampling scheme. They argued that α is distance and time-dependent at both the column and field scale, which complicates its spatial structure. Jacques et 523

al.(1999) found pure nugget effect in the variogram of K_f (Freundlich partition coefficient). Spatial structure of the Cl⁻ mass recovery was studied in a 2 m x 2 m x 2 m cube and a range of 0.37 m was found. With these results, we can argue that the range of the studied variables may vary depending on the sampling scheme and on the size of the studied site.

- 529
- 530 **Table 5**
- 531

Since the correlation among variables may depend on the spatial structure, the variation of the correlation coefficient among variables with the spatial scale was quantified. Fig.3 (A to D) shows these results for the correlations between *In K*, *In a* (K⁺), *In a* (Cl⁻), *K_d* (K⁺), and *K_d* (Cl⁻) and the variables which explained their variability, according to the MLR results. As stated by Wackernagel (1995), if the codispersion among the variables is constant in space, the structure of correlation of the variables is not affected by spatial scale.

539 The correlation coefficient among In K and Ma (Fig.3 A) decreases until 2.2 m and from 540 then on presents a variation around zero, showing that for distances larger than 2.2 m 541 these variables are no longer correlated. The spatial correlation among In K and ne 542 (Fig.3 A) showed that until 1.1 m the relationship became stronger and negative, 543 changing completely the kind of relation between these variables since it is recognized 544 that the increase in *ne* favors the water flow in soils. After that, the values became more 545 positive (an expected relation) but the correlation weaker until 2.8 m, when the variation 546 remained near zero. Similar behavior was also verified for the relation between ρ_d and 547 *Ma* and between *ne* and *Ma* (Fig.3 A), but the correlations were not statistically 548 significant. Contrarily, the correlation coefficient between *In K* and ρ_d became weaker 549 and positive until 3.3 m and then the variables seem to be not related in space. The 550 relation between *ne* and ρ_d was around zero for all studied distances (Fig.3 A).

The spatial correlation between K_d (K⁺) and *CEC* (Fig.3 B) presented a fast decrease until 1 m and then these variables are no longer statistically significant. The spatial correlation among K_d (K⁺) and *Ma* and among *CEC* and *Ma* (Fig.3 B) was not significant even for the distance equal to zero, but these variables were identified as explanatory in the MLR, illustrating the importance of the spatial analyses for a better decision about the most relevant variables and their relations.

557 Until a distance of 1m, only a slight decrease (become more negative) was observed in 558 the spatial correlations among K_d (Cl⁻) and Me, Me and clay content, and CEC and Me 559 (Fig.3 C). Contrarily, the correlation between K_d (Cl⁻) and clay content (Fig.3 C) showed 560 a slight increase until 1m. From 1 m, those correlations increased slightly and ranging 561 near zero, except the correlation between CEC and Me, which showed an erratic 562 behavior that may be related to its poor spatial correlation. The correlation among K_d (Cl⁻ 563) and CEC (Fig.3 C) became weaker and negative until 2 m but was statistically 564 significant just until 1.5 m. No spatial correlation was obtained between CEC and clay 565 content, result different from that obtained by Jacques et al. (1999) who observed a 566 strong spatial correlation between CEC and clay content until 2.03 m. Statistically significant correlation was verified between In α (K⁺) and In α (Cl⁺) (Fig.2 D) until 1.5 m, 567 568 and from 2 m these variables were no longer correlated.

569

570 **Fig. 3**

571 **4.** Conclusions

In this study, the spatial correlation among soil properties (total porosity, effective porosity, cation exchange capacity, macroporosity, microporosity, mesoporosity, bulk density, silt, clay and sand content) and the variables of interest (hydraulic conductivity, partition coefficient and dispersivity of a reactive (K⁺) and for a nonreactive solute (Cl⁻) was studied at the scale of a few meters using a dense sampling design. The soil was characterized as acid with low cation exchange capacity and composed of minerals commons for lateritic soils.

579 None of the variables studied were normally distributed. Soil properties presented a moderate coefficient of variation (CV) while hydraulic conductivity and transport 580 581 parameters were widely dispersed. None of the physical soil property was strongly 582 correlated to the transport parameters. Nevertheless, some parameters such as cation 583 exchange capacity and partition coefficient presented exhibits a statistically significant 584 positive correlation with transport parameters. Stepwise multiple linear regression (MLR) 585 analysis indicated that further studies should be performed aiming to include other 586 explanatory variables such as pH, electrical conductivity, the content of Al and Fe, 587 CaCO₃ and soil structure and microstructure, that are relevant variables for lateritic soils.

588 Our findings show that the use of geostatistical methods was efficient to evaluate the 589 spatial variation in the correlation coefficients. However, for the conditions analyzed, the 590 use of the spatial correlation among transport parameters and soil properties would 591 probably improve the estimation only in a small-scale study, since the spatial correlation

were only observed up to 2.5 m. It is important to mention that the study was performed for a specific field site and the results obtained may explain the spatial relation to the studied soil. However, the application of the statistical parameters to estimate transport parameters and predict solute transport in other soil is thus questionable.

The present study was a first attempt to evaluate the spatial correlation of transport parameters of a reactive and a nonreactive solute. We showed the soil properties that may exert greater influence and suggested the one that should be included in future studies. Understanding the spatial relations between variables can be useful in perform reliable prediction of flow and solute transport and contribute to reducing uncertainties when studying groundwater contamination.

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952 Fig. 1 Location of the study site and the position of the sampling points

953

- 954 Fig. 2 Column experiments in progress: a) water deionizers, b) hydraulic head
- 955 controller device, c) rigid-wall permeameters

956

957 Fig. 3 Variation of the correlation coefficient among variables with the spatial958 scale

960 Figure 1



Figure 2



Figure 3



| 970 List of Tables | |
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972 Table 1 Descriptive statistics of soil properties, hydraulic conductivity and linear

973 average velocity at the study site

Table 2 Descriptive statistics of transport parameters

Table 3 Correlation coefficients among standardized variables

Table 4 Stepwise multiple linear regression results

Table 5 Parameters of the variogram models used to fit the isotropic direct
 experimental variograms

986 **Table 1**

| | Mean | SD | CV | Skew | Kurt | Min | Max |
|---|-------|------|------|-------|-------|-------|-------|
| <i>K</i> [m d ⁻¹] | 1.35 | 1.65 | 1.22 | 2.39 | 5.84 | 0.03 | 7.46 |
| <i>v</i> [m d ⁻¹] | 5.40 | 6.57 | 1.23 | 2.20 | 4.61 | 0.13 | 27.71 |
| <i>In K</i> [In (m d ⁻¹)] | -0.37 | 1.25 | n.d | -0.29 | -0.12 | -3.68 | 2.03 |
| n [] | 0.51 | 0.04 | 0.08 | -0.24 | -0.39 | 0.42 | 0.58 |
| ne[] | 0.24 | 0.02 | 0.08 | -0.39 | -0.14 | 0.20 | 0.28 |
| ρ _d [g cm ⁻³] | 1.34 | 0.10 | 0.07 | 0.28 | -0.32 | 1.14 | 1.59 |
| CEC [cmol _c Kg ⁻¹] | 2.51 | 0.64 | 0.25 | 1.06 | 0.39 | 1.60 | 4.20 |
| sand (%) | 56.20 | 3.24 | 0.06 | -0.36 | -0.52 | 48.50 | 61.50 |
| silt (%) | 4.62 | 2.82 | 0.61 | 0.16 | -0.06 | 1.40 | 11.40 |
| clay (%) | 39.18 | 3.51 | 0.09 | 0.10 | -0.87 | 32.50 | 46.10 |
| Ma [] | 0.072 | 0.04 | 0.56 | 0.75 | -0.58 | 0.031 | 0.152 |
| Mi [] | 0.262 | 0.06 | 0.23 | -0.25 | -0.96 | 0.141 | 0.361 |
| Me[] | 0.172 | 0.05 | 0.29 | 0.21 | -0.92 | 0.091 | 0.263 |
| | | | | | | | |

987 SD: standard deviation, CV: coefficient of variation, Skew: Skewness, Kurt: Kurtosis, 988 Min: minimum value, Max: maximum value, n.d: undetermined, *K*: hydraulic conductivity; 989 *v*: linear average velocity, ρ_d : bulk density, *n*: total porosity, *ne*: effective porosity *Ma*: 990 macroporosity, *Me*: mesoporosity, *Mi*: microporosity, *CEC*: cation exchange capacity. 991

| 992 | Table 2 |
|-----|---------|
| /// | |

| | Mean | SD | CV | Skew | Kurt | Min | Max |
|---|-------|------|------|-------|-------|-------|-------|
| Р(К+)[] | 2.07 | 2.12 | 1.02 | 2.15 | 5.61 | 0.11 | 10.80 |
| <i>R</i> (K ⁺) [] | 5.37 | 5.10 | 0.95 | 4.51 | 25.31 | 0.69 | 36.19 |
| <i>Kd</i> (K ⁺) [cm ³ g ⁻¹] | 1.71 | 2.27 | 1.33 | 5.61 | 36.75 | 0.01 | 16.75 |
| <i>D</i> (K⁺) [m²d⁻¹] | 1.07 | 1.77 | 1.65 | 2.64 | 7.42 | 0.02 | 8.77 |
| α (K⁺) [m] | 0.18 | 0.19 | 1.06 | 1.82 | 3.32 | 0.01 | 0.88 |
| <i>ln</i> α (K⁺) [ln (m)] | -2.21 | 1.11 | n.d | -0.45 | 0.71 | -5.79 | -0.12 |
| P (Cl ⁻) [] | 2.82 | 2.78 | 0.99 | 2.08 | 4.25 | 0.44 | 13.41 |
| <i>R</i> (Cl ⁻) [] | 2.35 | 1.29 | 0.55 | 0.61 | -0.60 | 0.33 | 5.20 |
| <i>Kd</i> (Cl ⁻) [cm ³ g ₋₁] | 0.55 | 0.51 | 0.93 | 0.66 | -0.81 | 0.03 | 1.64 |
| D (Cl⁻) [m²d⁻¹] | 0.61 | 1.14 | 1.87 | 3.43 | 12.26 | 0.01 | 5.62 |
| α (Cl ⁻) [m] | 0.10 | 0.08 | 0.80 | 1.23 | 1.43 | 0.01 | 0.34 |
| <i>ln α</i> (Cl ⁻) [ln (m)] | -2.61 | 0.93 | n.d | -1.18 | 2.80 | -6.18 | -1.07 |

994 SD: standard deviation, CV: coefficient of variation, Skew: Skewness, Kurt: Kurtosis, 995 Min: minimum value, Max: maximum value, n.d.: undetermined, *P*: Peclet number, *R*: 996 retardation coefficient, K_d : partition coefficient, *D*: hydrodynamic dispersion coefficient, *a*: 997 dispersivity, (K⁺) potassium, (Cl⁻): chloride.

| | In α (Cl ⁻) | In α (K ⁺) | $K_d(K^*)$ | K _d (Cl⁻) | CEC | InK | $ ho_d$ | n | ne | sand | silt | clay | Mi | Ме | Ma |
|-------------------------------|-------------------------|---------------------------|------------|--------------------------|-------|--------|---------|-------|-------|---------|---------|-------|---------|---------|------|
| In α (Cl ⁻) | 1.00 | | | | | | | | | | | | | | |
| <i>ln</i> α (K ⁺) | 0.71** | 1.00 | | | | | | | | | | | | | |
| K_d (K ⁺) | -0.03 | 0.06 | 1.00 | | | | | | | | | | | | |
| K _d (Cl⁻) | -0.09 | 0.10 | 0.63** | 1.00 | | | | | | | | | | | |
| CEC | -0.04 | 0.15 | 0.70** | 0.81** | 1.00 | | | | | | | | | | |
| In K | 0.13 | 0.11 | -0.02 | -0.14 | -0.10 | 1.00 | | | | | | | | | |
| ${oldsymbol{ ho}_d}$ | 0.33* | 0.11 | -0.22 | -0.20 | -0.26 | -0.34* | 1.00 | | | | | | | | |
| n | -0.32* | -0.10 | 0.23 | 0.19 | 0.26 | 0.33* | -0.99** | 1.00 | | | | | | | |
| ne | 0.25 | 0.12 | 0.06 | 0.02 | -0.01 | 0.44** | 0.06 | -0.06 | 1.00 | | | | | | |
| sand | 0.21 | 0.18 | -0.27 | -0.13 | -0.17 | 0.01 | 0.22 | -0.22 | -0.28 | 1.00 | | | | | |
| silt | -0.07 | -0.10 | 0.16 | 0.30* | 0.21 | 0.05 | -0.10 | 0.08 | 0.11 | -0.29* | 1.00 | | | | |
| clay | -0.14 | -0.10 | 0.15 | -0.09 | 0.01 | -0.05 | -0.11 | 0.13 | 0.16 | -0.69** | -0.48** | 1.00 | | | |
| Mi | -0.21 | -0.18 | -0.39** | -0.25 | -0.26 | -0.02 | 0.04 | -0.03 | -0.22 | 0.10 | -0.37** | 0.18 | 1.00 | | |
| Me | 0.08 | 0.06 | 0.36* | 0.28 | 0.12 | -0.19 | 0.18 | -0.20 | 0.22 | -0.15 | 0.40** | -0.14 | -0.68** | 1.00 | |
| Ма | 0.10 | 0.03 | -0.16 | -0.20 | 0.03 | 0.38** | -0.22 | 0.24 | 0.10 | 0.14 | -0.02 | -0.12 | 0.28 | -0.63** | 1.00 |

 α : dispersivity, (K⁺) potassium, (Cl⁻): chloride, K_d : partition coefficient, CEC: cation exchange capacity, K:

hydraulic conductivity; ρ_d : bulk density, *n*: total porosity, *ne*: effective porosity, *Mi*: microporosity, *Me*: mesoporosity, *Ma*: macroporosity * significant at 0.01 level of significance. 1003

1006 ** significant at 0.05 level of significance.

| 1007 | Table 4 |
|------|---------|
| | |

| 1007 | |
|------|--|
| 1008 | |
| 1000 | |

| | $K_d(K^+)$ | <i>In</i> α (K ⁺) | K _d (Cl⁻) | In α (Cl ⁻) | In K |
|-------------------------|------------|-------------------------------|----------------------|-------------------------|-------|
| In K | - | - | - | - | |
| n | - | - | - | - | - |
| ne | - | - | - | - | 26.4 |
| $ ho_{d}$ | - | - | - | - | -3.90 |
| CEC | 0.68 | - | 0.48 | - | - |
| sand | - | - | - | - | - |
| silt | - | - | - | - | - |
| clay | - | - | 0.03 | - | - |
| Ма | -0.03 | - | - | - | 0.09 |
| Mi | - | - | - | - | - |
| Me | - | - | 0.02 | - | - |
| $K_d(K^*)$ | | - | - | - | - |
| <i>ln</i> α (K⁺) | - | | - | 0.63 | - |
| K _d (Cl⁻) | - | - | | - | - |
| In α (Cl ⁻) | - | 0.79 | - | | - |
| Intercept | -0.29 | -0.26 | -2.34 | -1.15 | -2.25 |
| R ² | 0.70 | 0.50 | 0.60 | 0.50 | 0.40 |
| r* | 0.84 | 0.70 | 0.77 | 0.72 | 0.63 |
| | | | | | |

* Pearson's coefficient

- All results were significant at p =< 0.05
- 1012 1013 1014

Table 5

| | Variable | Madal | Nugget | | Range (m) | |
|----------|-------------------------------|--------------------|------------------------|------|-----------|--|
| Variable | Model | (C ₀) | Siii (C ₁) | (a) | | |
| | In K | Spherical | 0.0 | 1.0 | 4.0 | |
| | n | Spherical | 0.0 | 1.0 | 3.0 | |
| | ne | Spherical | 0.0 | 1.0 | 3.0 | |
| | $ ho_{d}$ | Spherical | 0.0 | 1.0 | 3.5 | |
| | CEC | Spherical | 0.0 | 1.0 | 3.0 | |
| | sand | Spherical | 0.0 | 1.0 | 3.0 | |
| | silt | Spherical | 0.0 | 1.0 | 2.5 | |
| | clay | Pure nugget effect | 1.0 | 0.0 | - | |
| | Ма | Spherical | 0.0 | 1.0 | 3.5 | |
| | Mi | Spherical | 0.45 | 0.55 | 2.5 | |
| | Ме | Spherical | 0.00 | 1.0 | 3.0 | |
| | K_d (K ⁺) | Spherical | 0.40 | 0.60 | 3.6 | |
| | <i>ln</i> α (K ⁺) | Spherical | 0.50 | 0.50 | 3.0 | |
| | K_d (CI ⁻) | Spherical | 0.55 | 0.45 | 3.3 | |
| | ln α (Cl⁻) | Spherical | 0.30 | 0.70 | 2.7 | |
| | | | | | | |